

B.Sc. Biotechnology: Semester-II
BST 203 Chemistry II

Teaching Scheme Lectures: 3 hrs/Week Tutorials: 1 hr/Week Credits: 4	Examination Scheme Class Test -12Marks Teachers Assessment - 6Marks Attendance – 12 Marks End Semester Exam – 70 marks
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Prerequisite: - General knowledge of Chemistry

Course Objectives:

- 1.To give an overview of Chemical kinetics
- 2.To give basic knowledge of chemicals and their reactions
- 3.To have an overview of mathematical characteristics of simple chemical reactions
- 4.Aromatic electrophilic substitution- general pattern of the mechanism,
- 5.Activating and deactivating substituents
- 5.To explain the complexation tendencies including their function in biosystems

Detail Syllabus

Module-1

Chemical kinetics and its scope, rate of a reaction, Order of the reactions. Concentration dependence of rates, mathematical characteristics of simple chemical reactions-zero order, first order, second order, pseudo order, half life and mean life.

Module-2

Aromatic electrophilic substitution- general pattern of the mechanism, Mechanism of nitration, halogenation, sulphonation, mercuration and Friedel-Crafts reaction, Activating and deactivating substituents, orientation and ortho/para ratio.

Module-3

s-Block Elements Comparative study, diagonal relationships, salient features of hydrides, solvation and complexation tendencies including their function in biosystems, and introduction to alkyls and aryls.

p-Block Elements Comparative study (including diagonal relationship) of groups 13-17 elements, compounds like hydrides, oxides and halides of groups 13-16, hydrides of boron diborane and higher boranes, borazine, borohydrides.

Text and Reference Books

- 1.A Textbook of Physical Chemistry, A. S. Negi, S. C. Anand
- 2.Physical Chemistry, Gilbert William Castellan
- 3.Physical chemistry, Walter John Moore
- 4.Organic Chemistry, Benjamin List, Keiji Maruoka
- 5.Advanced Organic Chemistry, 4th ed. Part A: Structure and Mechanisms F. Carey and R. Sundberg, Kluwer Academic



Head

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